

Precipitation Systems and Microphysical Processes

Atmos 5210: Synoptic Meteorology II



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Precipitation Systems

Generating Clouds and Precipitation

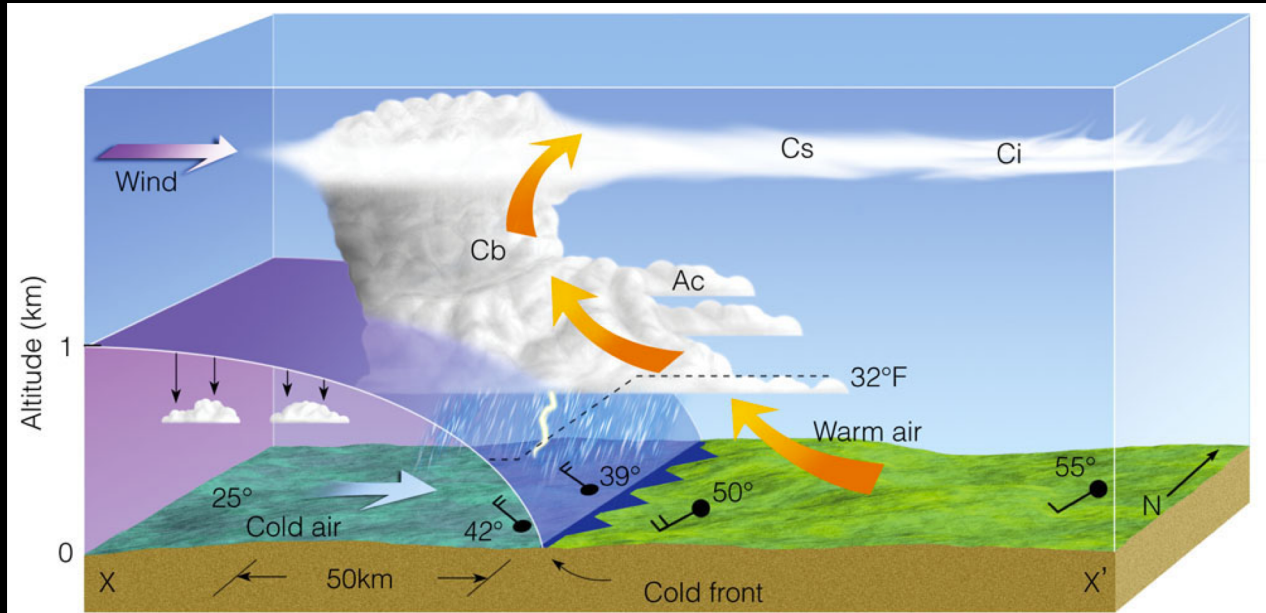
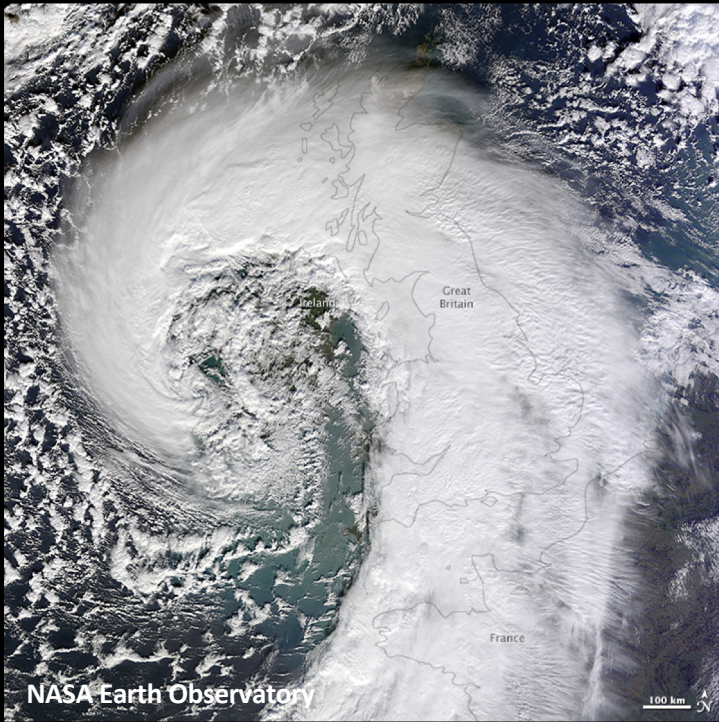
- Clouds form when water vapor in the atmosphere condenses into cloud droplets or ice crystals
 - Requires air to become supersaturated through evaporation or cooling
 - Ascent and associated adiabatic expansion and cooling is the primary (but not only) mechanism for generating supersaturation in precipitating clouds
- Precipitation occurs when hydrometeors grow sufficiently large to fall and reach the ground
 - Typically cannot be accomplished through condensation alone
 - May involve multiple microphysical processes

Group Discussion

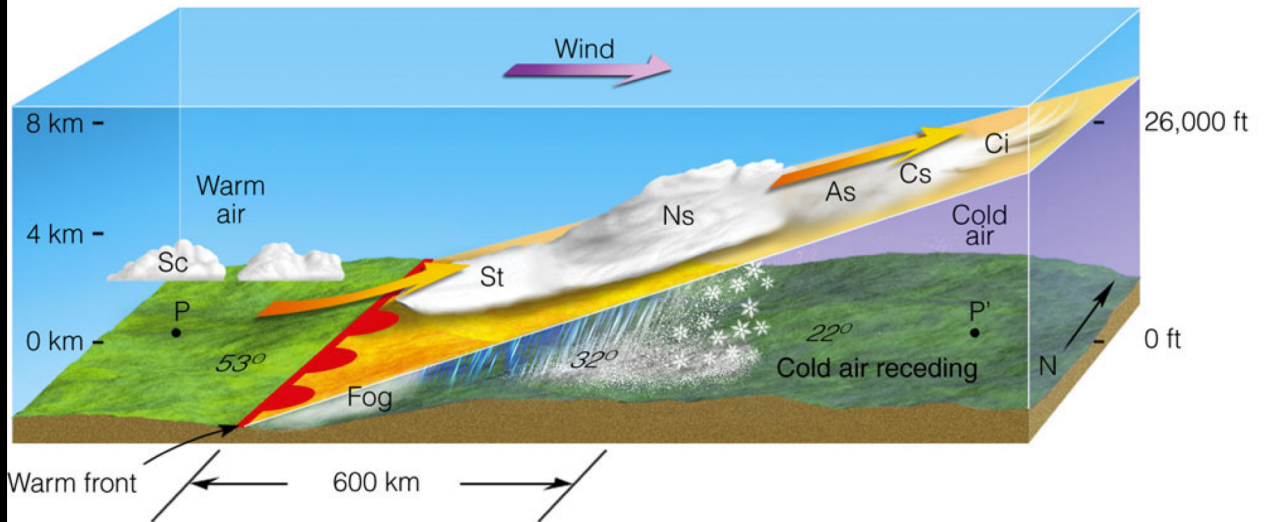
What weather systems are primarily responsible for precipitation in the tropics and extratropics?



Extratropical Cyclones/Fronts



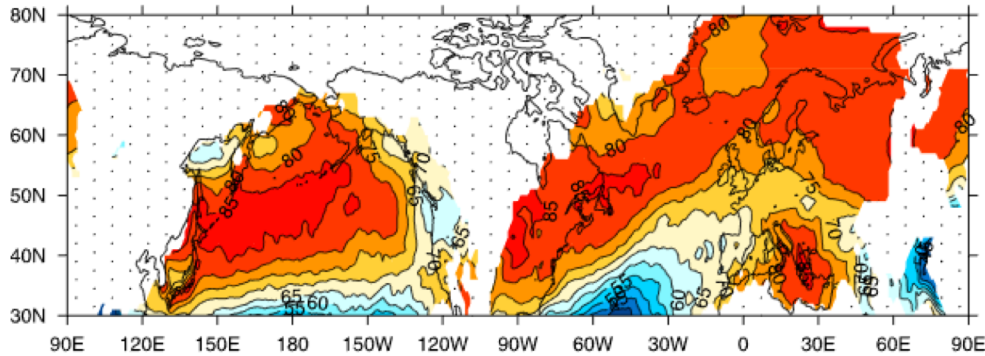
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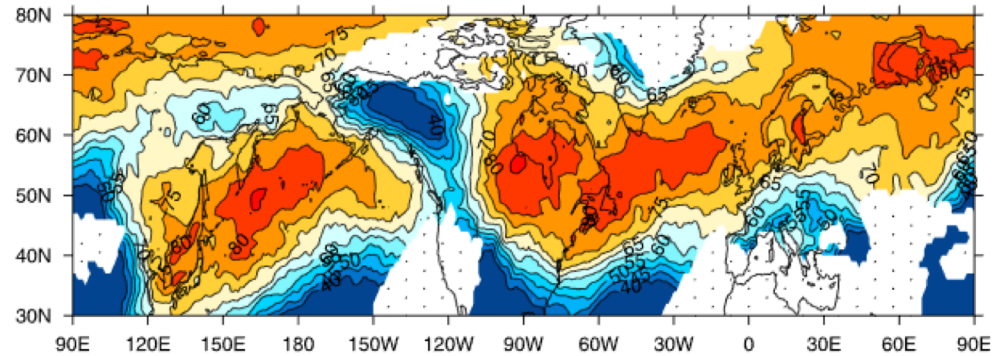
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Cyclone Contribution

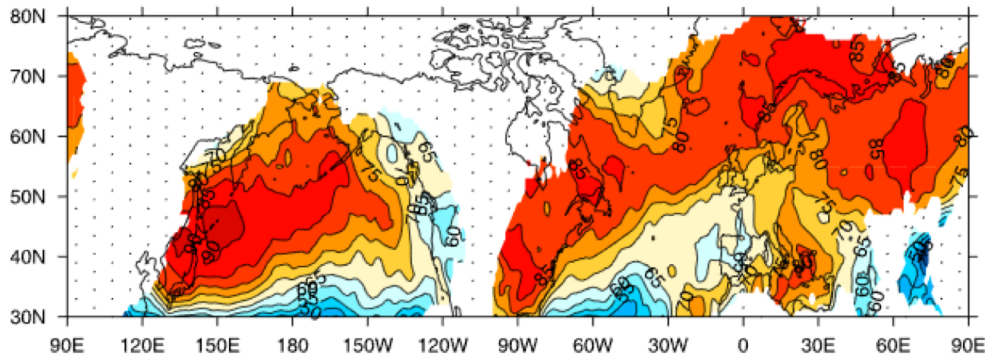
ERA-Interim DJF



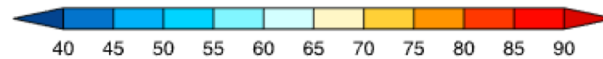
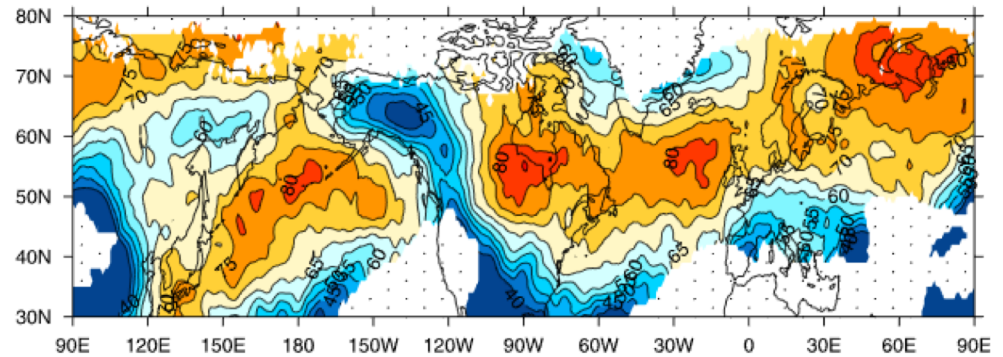
ERA-Interim JJA



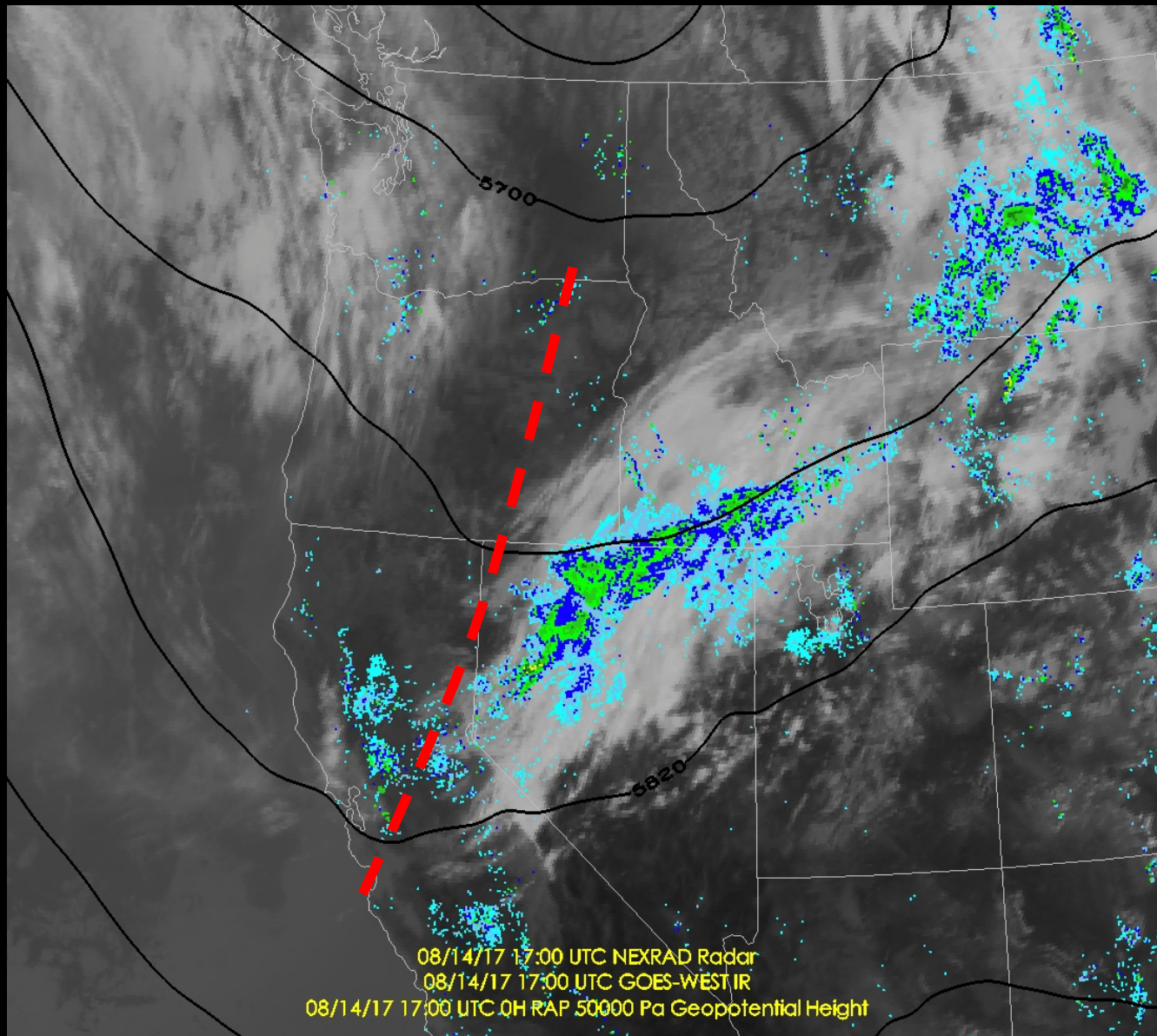
GPCP DJF



GPCP JJA



Large-scale forcing



Convection

- Vertical motions due to an imbalance of forces in the vertical
- Precipitating clouds generated by rising parcels that are warmer than their environment, resulting in an updraft
- Key ingredients
 - Instability, moisture, & lift



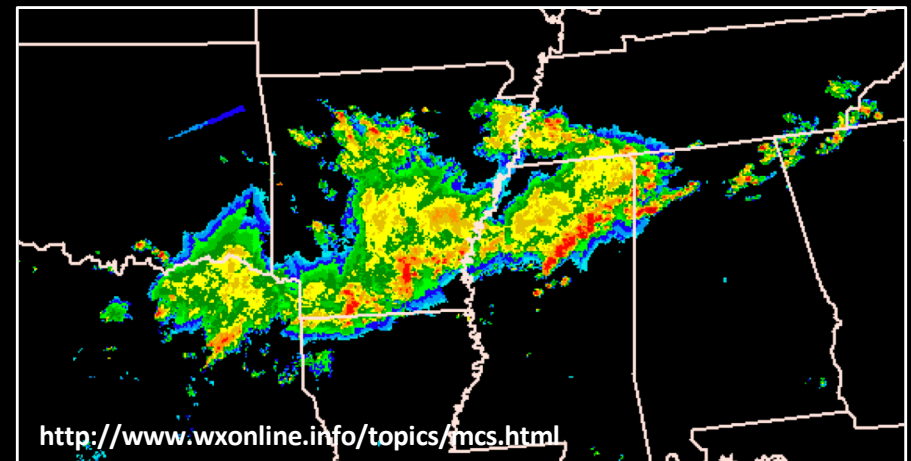
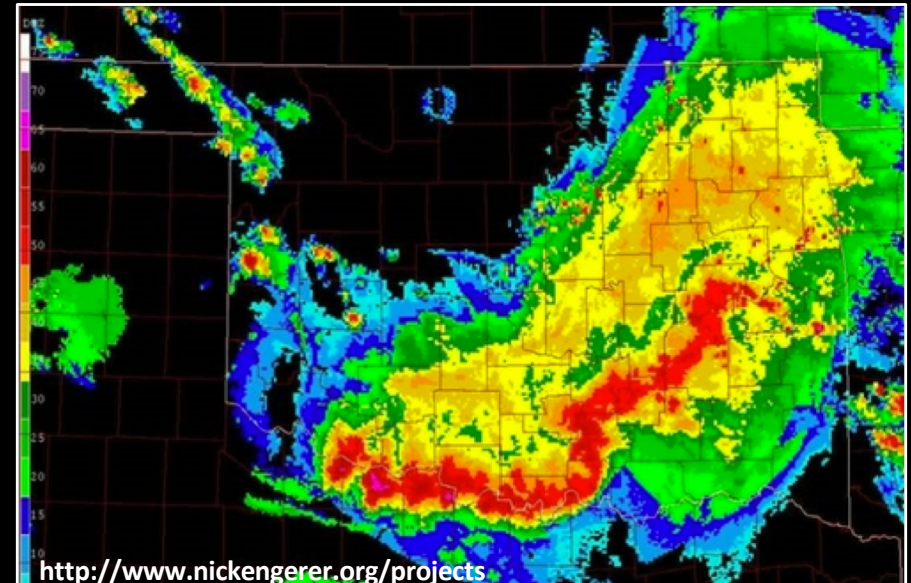
Convection



<https://www.youtube.com/watch?v=Py9thuWwofl>

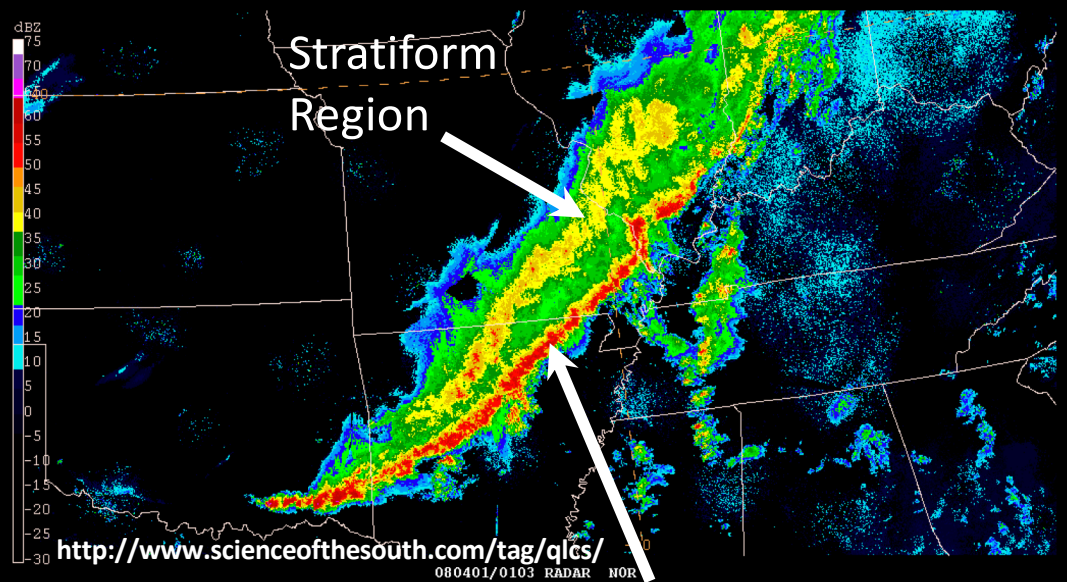
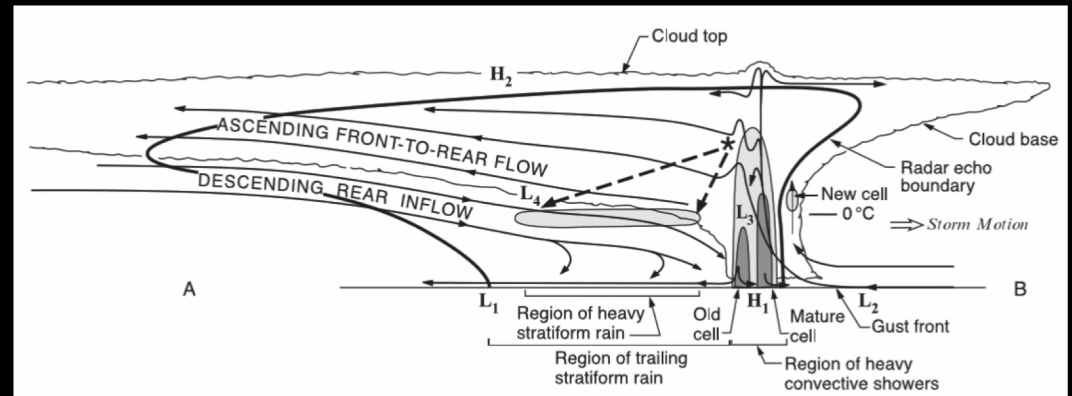
Mesoscale Convective Systems

- Organized collection of two or more cumulonimbus clouds that interact to form an *extensive* region of precipitation
- Precipitation region is nearly contiguous and contains convective and stratiform elements, with the latter typically more extensive

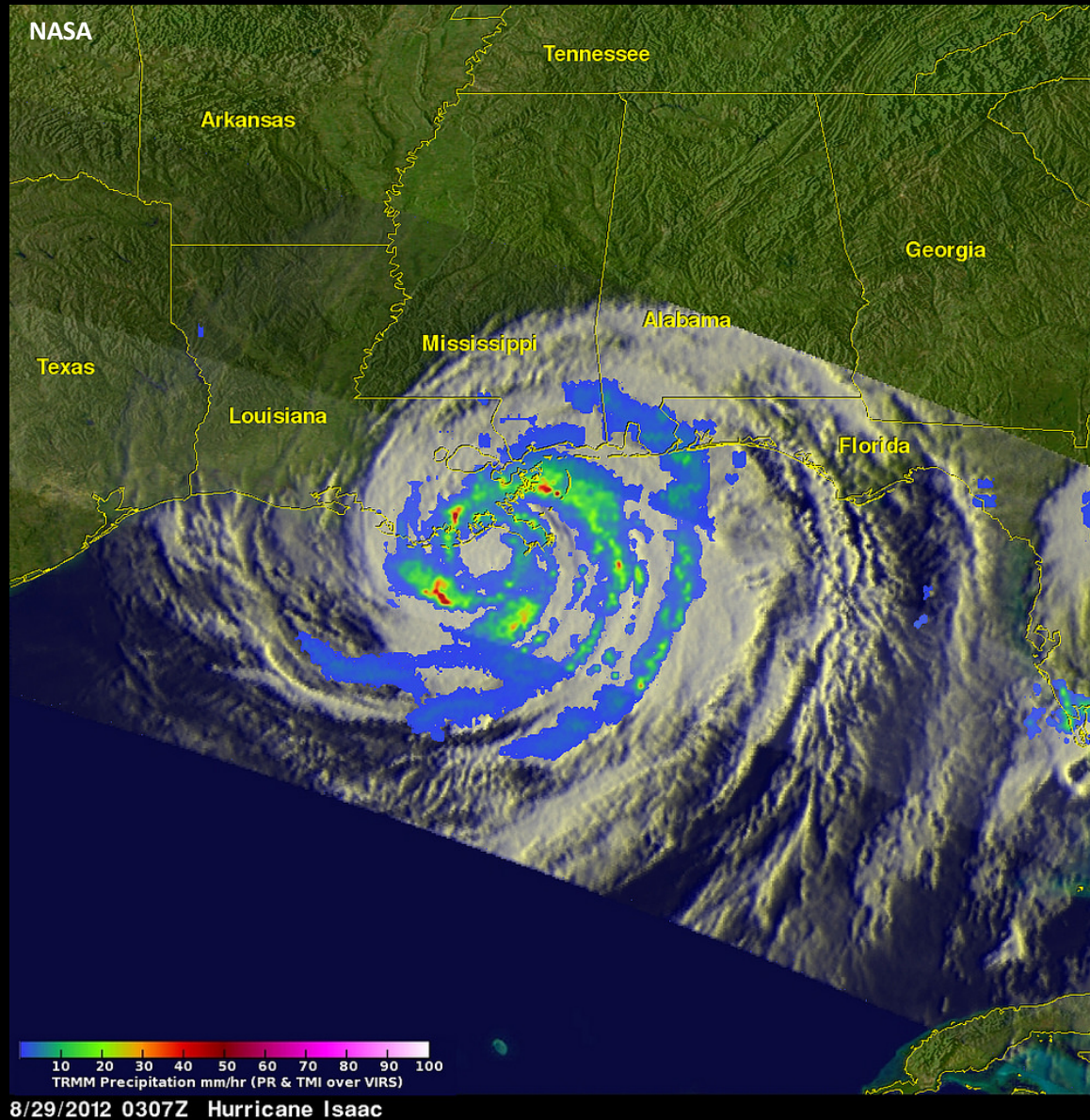


Mesoscale Convective Systems

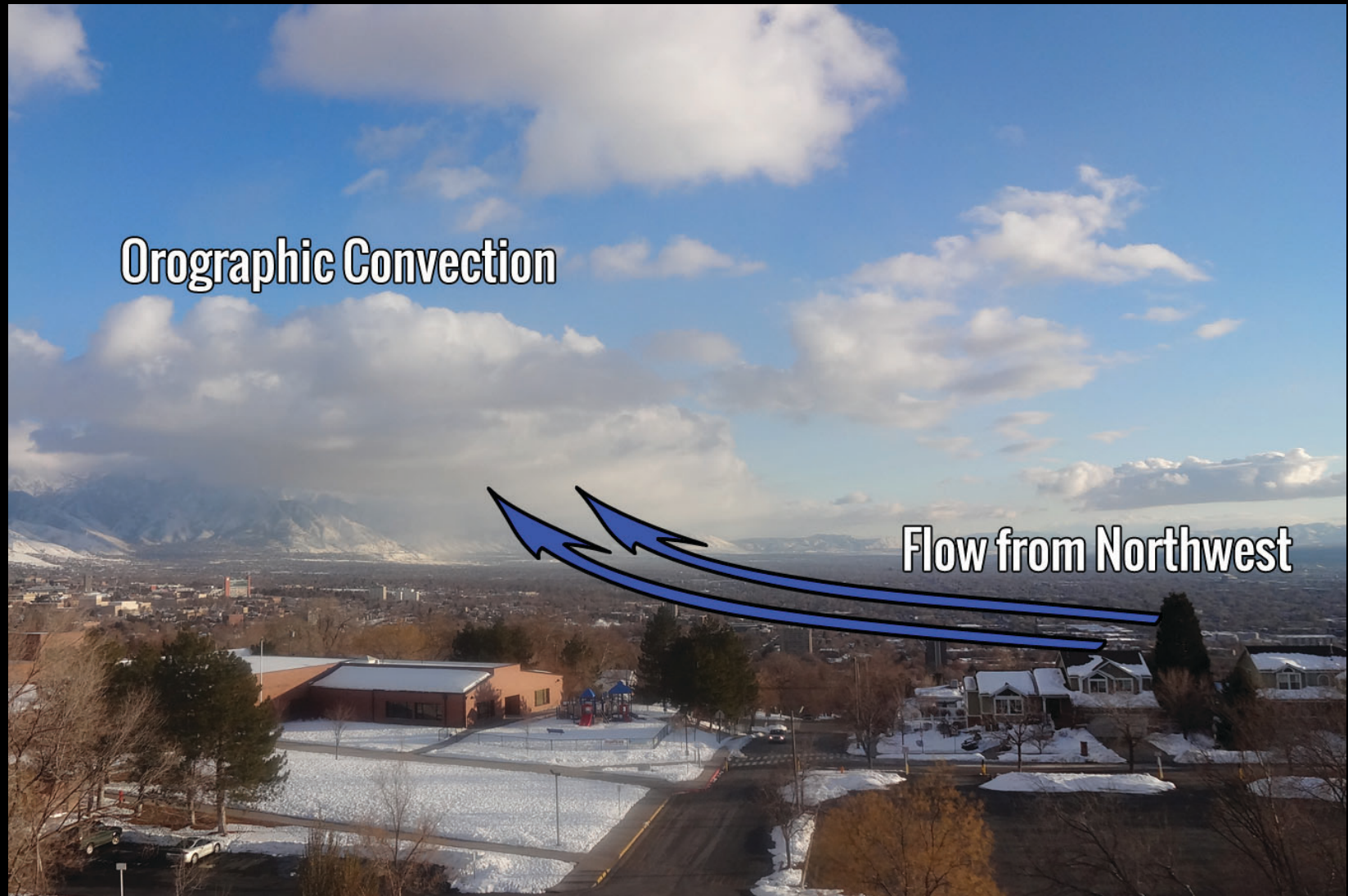
- Hydrometeor detrainment and transport from convective line and “layer lifting” important in precipitation dynamics of the stratiform region
- Example is of a convective-line MCS (a.k.a. squall line)



Tropical Cyclones



Orographic

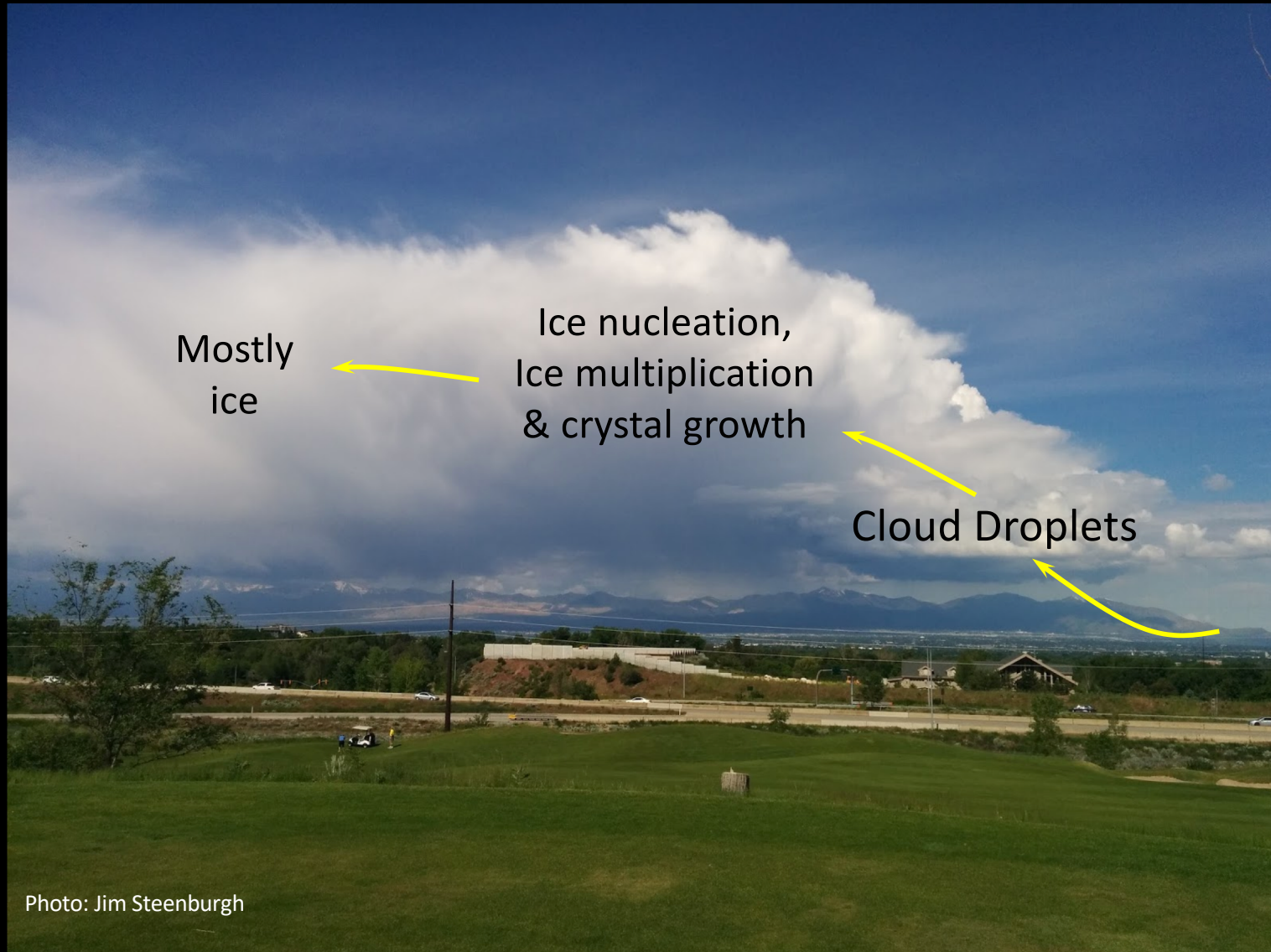


Microphysical Processes

Microphysical Processes

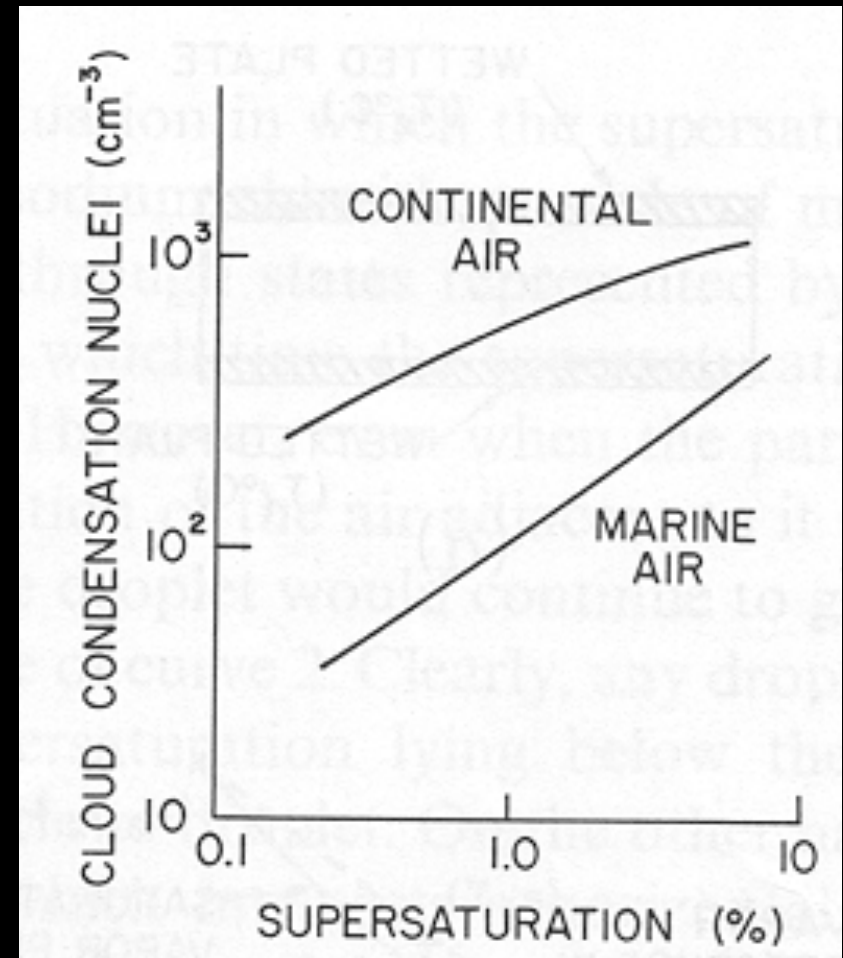
- Cloud droplet formation
 - CCN and droplet size spectra
- Warm cloud processes
 - Collision-coalescence
- Mixed-phase processes
 - Ice nucleation
 - Ice multiplication
 - Depositional growth (a.k.a., the Bergeron-Findeisen Process)
 - Accretional growth
 - Aggregation

Microphysical Processes



Droplet Formation

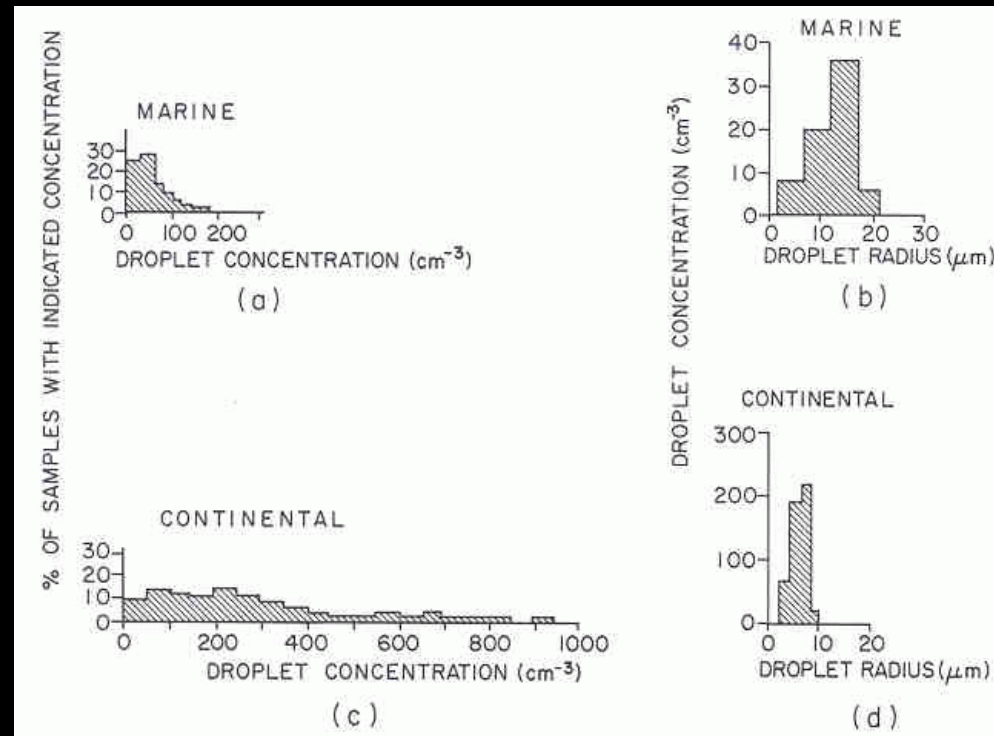
- Cloud droplets form on soluble atmospheric aerosols
 - *Heterogeneous nucleation*
- *Cloud condensation nuclei (CCN)*
 - Aerosol which serve as nuclei for water vapor condensation
- On *average*, there is an order of magnitude more CCN in continental air than maritime air



Size Spectra

Marine

Continental

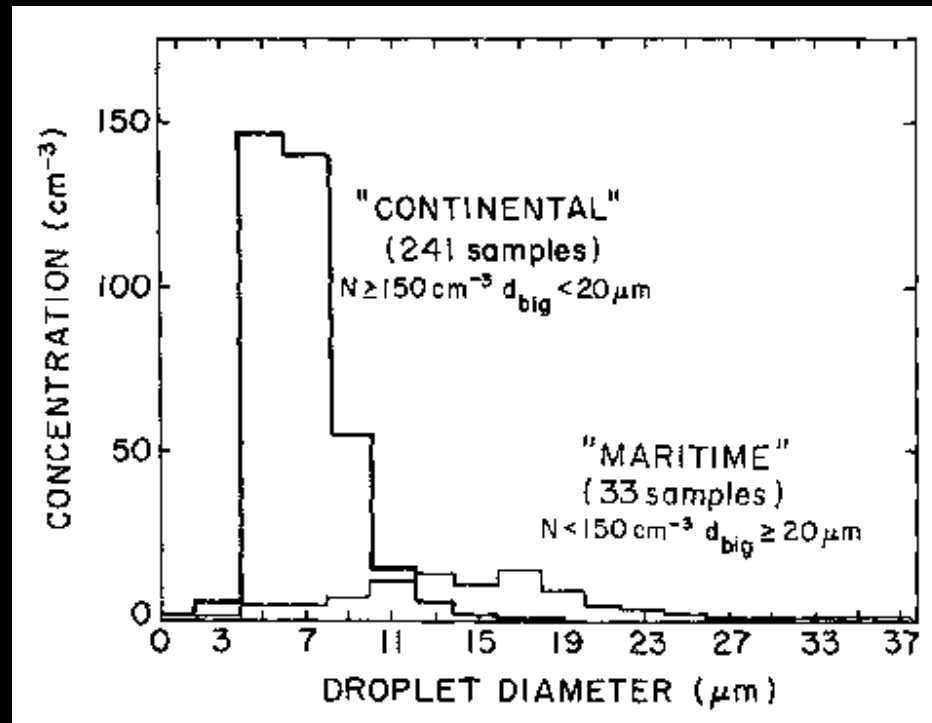


- Continental clouds frequently feature
 - Large cloud droplet number concentrations & smaller cloud droplets
- Maritime clouds frequently feature
 - Smaller cloud droplet number concentrations & larger cloud droplets

Size Spectra

Cloud droplet spectra

Storm Peak Lab
Steamboat, CO

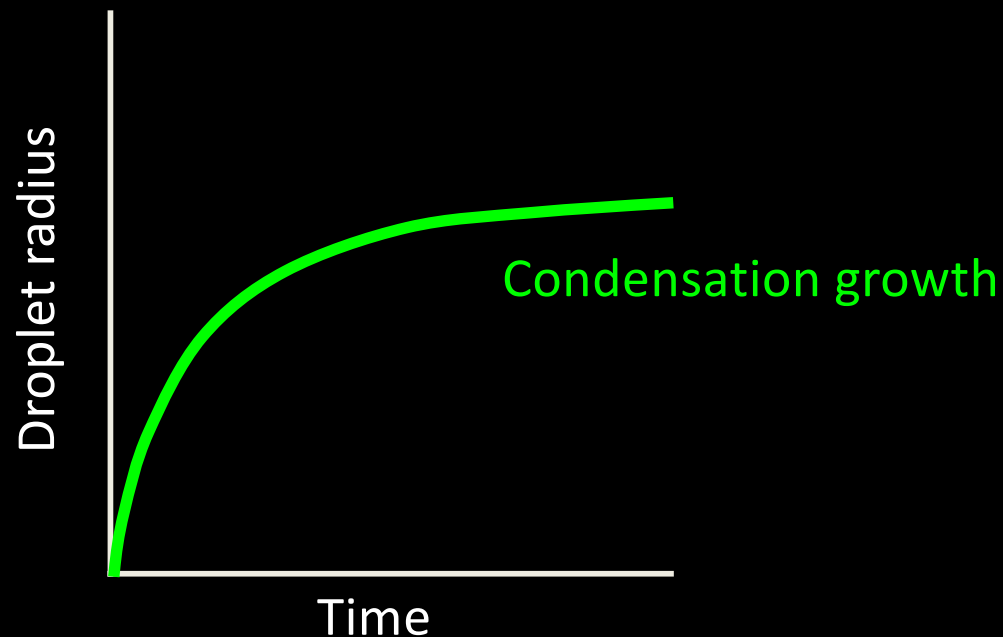


- In continental areas, however, there are large intra- and inter-storm variations depending on aerosol characteristics
 - Maritime size spectra are rare, but possible
- Significance: Impacts hydrometeor growth (more later)

Warm Cloud Processes

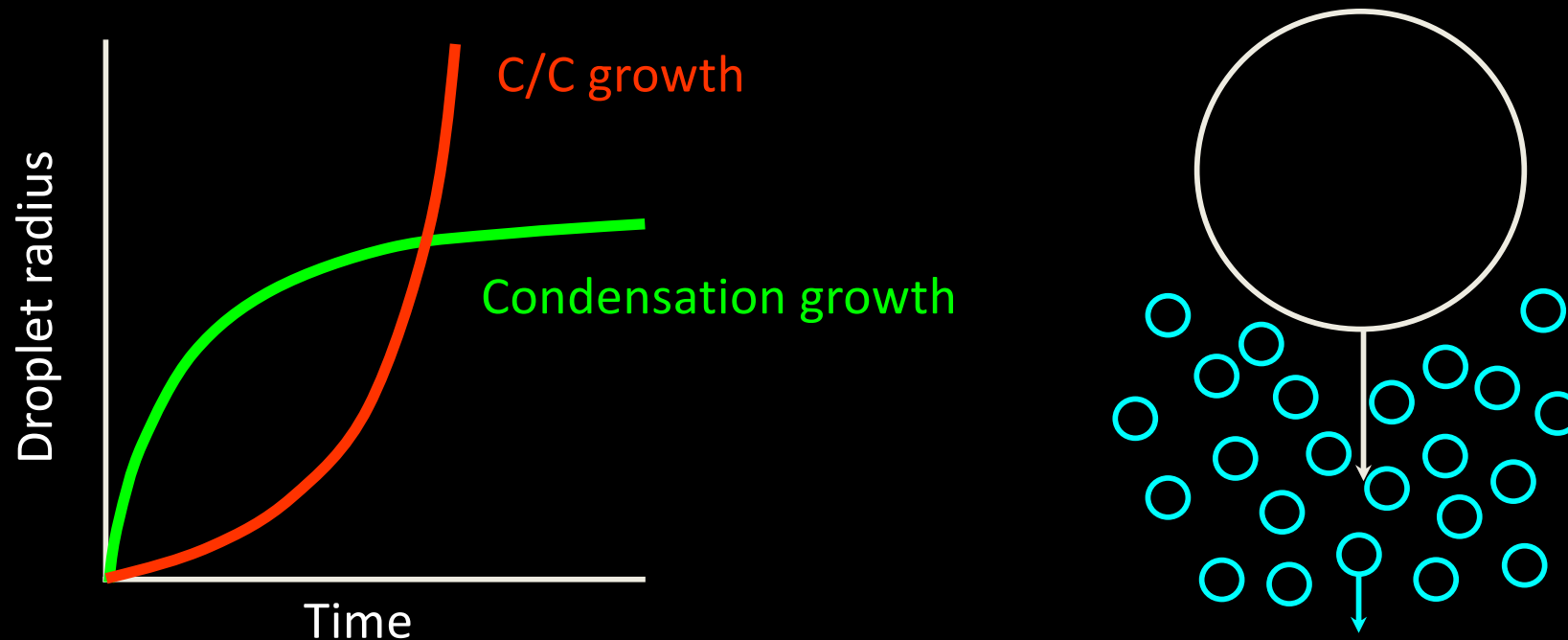
- “Warm Cloud”
 - Clouds that lie entirely below the 0°C level or consist entirely of liquid water
- Mechanisms for warm cloud hydrometeor growth
 - Condensation
 - Collision-coalescence

Condensation



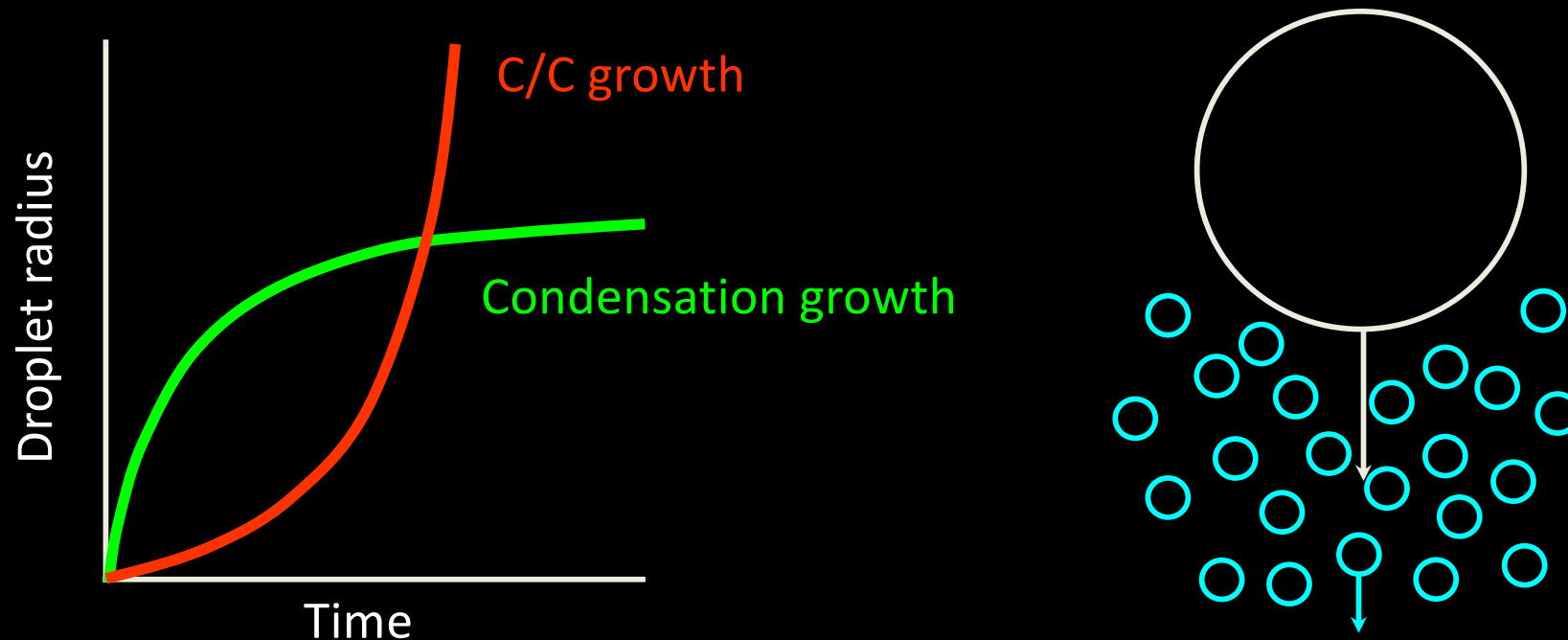
- Droplet growth by condensation is initially rapid, but slows with time
- Condensational growth too slow to produce large raindrops

Collision–Coalescence



- Growth of small droplets into raindrops is achieved by *collision-coalescence*
- Fall velocity of droplet increases with size
- Larger particles sweep out smaller cloud droplets and grow
- Becomes more efficient as radius increases
- Turbulence may contribute to this growth mechanism

Warm Cloud Processes



- Cloud droplet growth initially dominated by condensation
- Growth into raindrops dominated by collision-coalescence
-
- Most effective in maritime clouds due to presence of larger cloud droplets (due to fewer CCN)

Hawaiian Warm-Phase Clouds

Freezing Level (0°C)

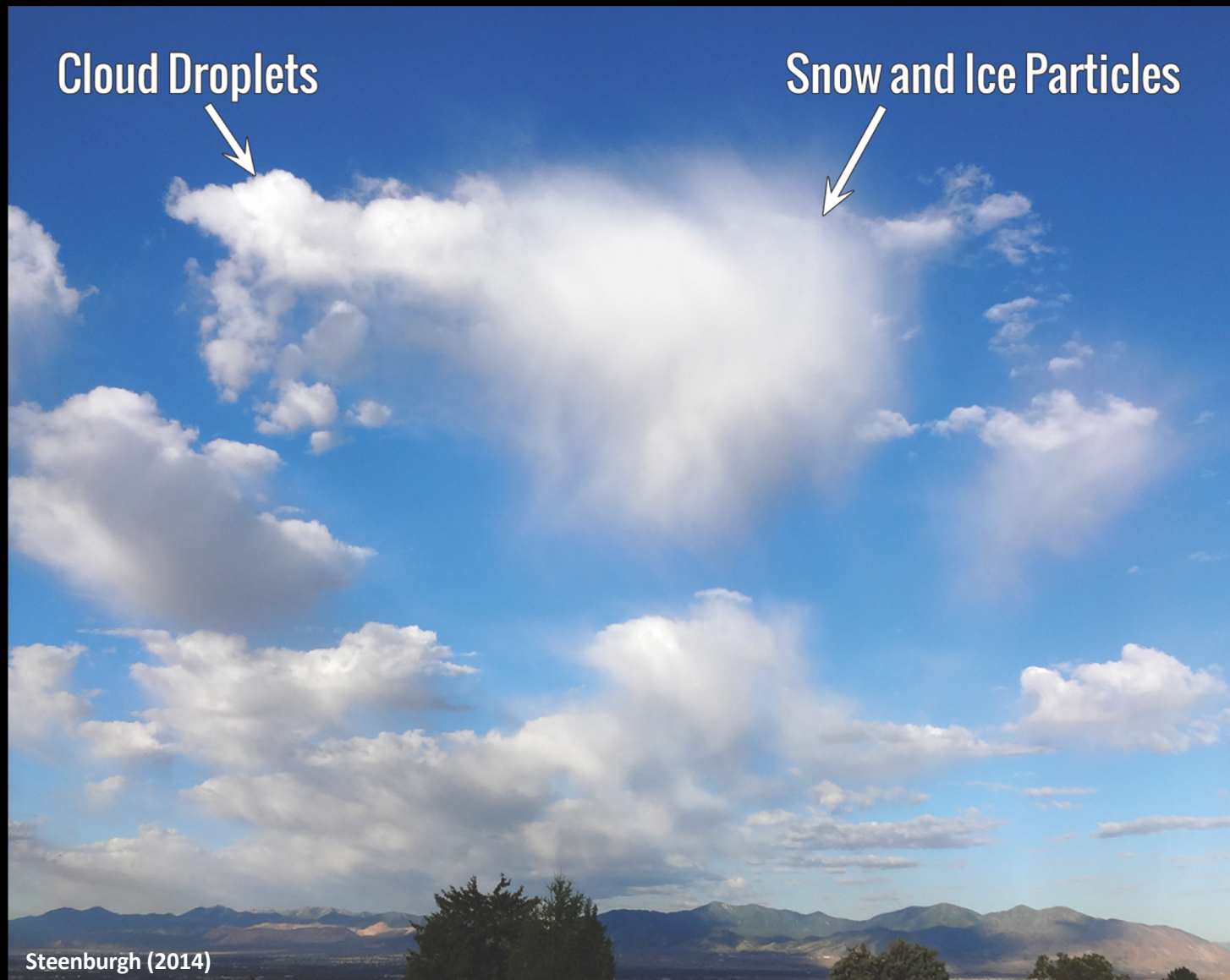
Liquid water cloud droplets
&
Collision-coalescence

Rain

Mixed-Phase Cloud Processes

- Glaciation
 - Ice nucleation & multiplication
- Depositional growth
- Accretion
- Aggregation

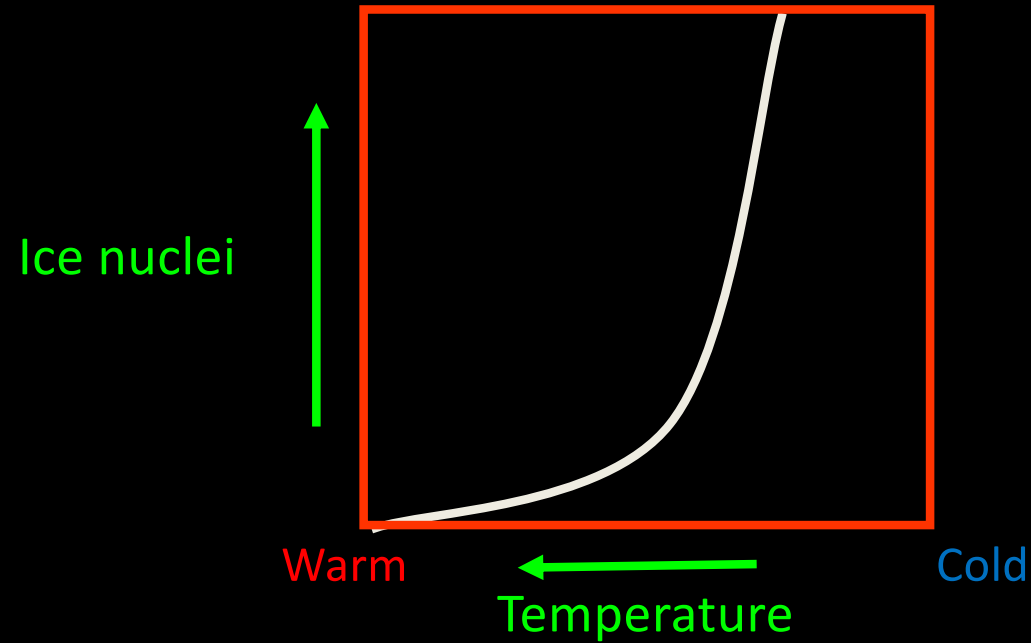
Glaciating Cloud Example



Ice Nucleation

- Water does not freeze at 0°C
 - Pure water does not freeze until almost -40°C (homogeneous nucleation)
 - Supercooled liquid water (SLW) – water (rain or cloud droplets) that exists at temperatures below 0°C
 - Ice nuclei – enable water to freeze at temperatures above -40°C
- The effectiveness of potential ice nuclei is dependent on
 - Molecular spacing and crystal structure - similar to ice is best
 - Temperature – Activation is more likely as temperature decreases
- Ice nuclei concentration increases as temperature decreases

Ice Nucleation



Ice nuclei increase by an order of magnitude for every 4°C drop in temperature

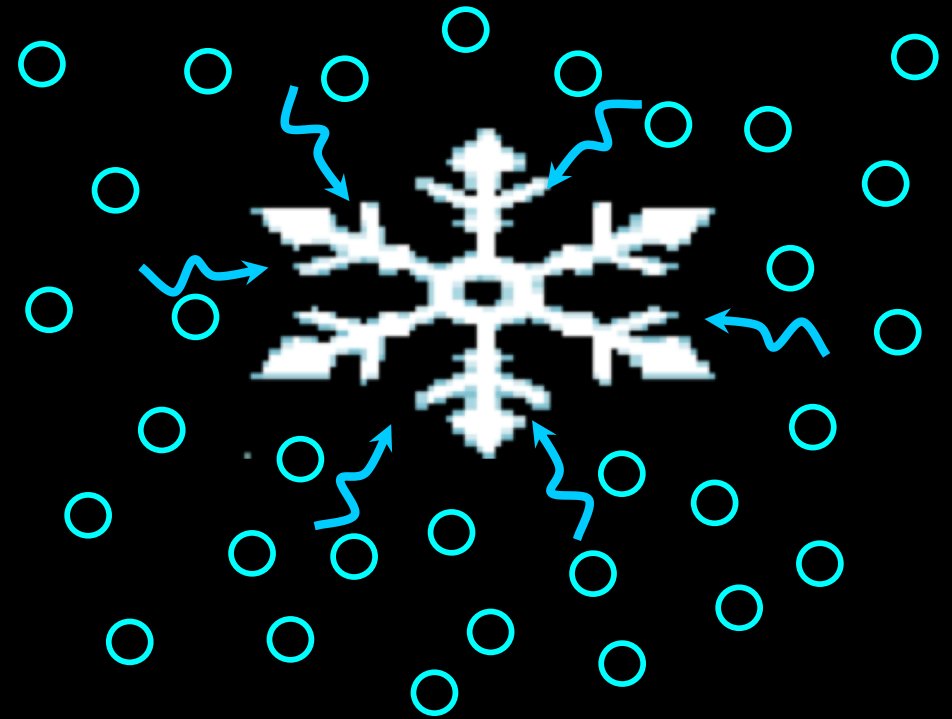
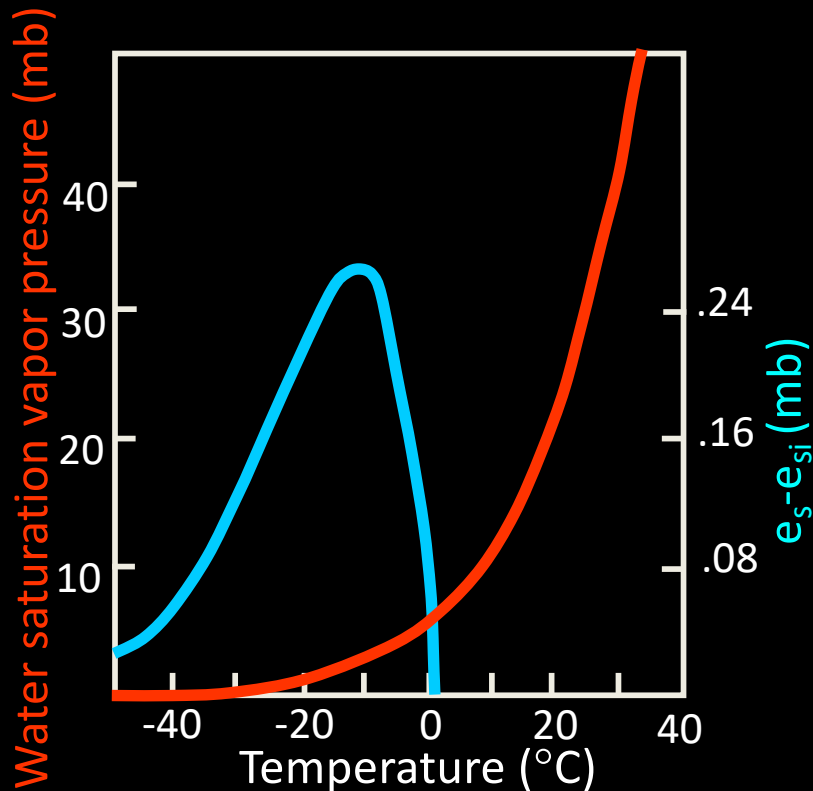
Ice Nucleation

- Significance?
 - Cloud will not necessarily glaciate at temperatures below 0°C
 - Want snow (or even rain in many cases)? Need ice!
 - If temperatures in cloud are
 - -4°C or warmer VERY LITTLE chance of ice
 - -10°C 60% chance of ice
 - -12°C 70% chance of ice
 - -15°C 90% chance of ice
 - 20°C VERY GOOD chance of ice

Ice Multiplication

- Still have a few problems
 - There are still very few ice nuclei even at cold temperatures
 - Ice particle concentrations greatly exceed ice nuclei concentrations in most mixed phase clouds
 - How do we get so much ice?
- Ice multiplication – creation of large numbers of ice particles through
 - Mechanical fracturing of ice crystals during evaporation
 - Shattering of large drops during freezing
 - Splintering of ice during riming (Hallet-Mossop Process)

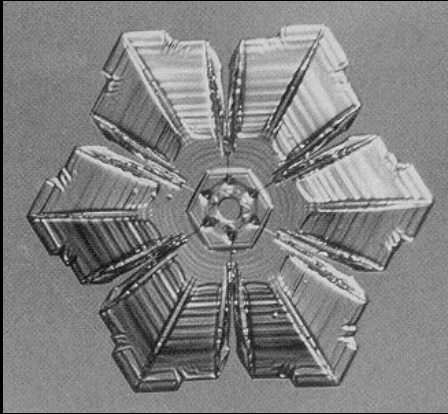
Deposition (WBF Process)



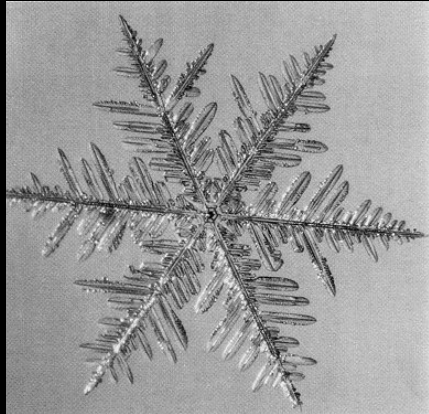
Wegener-Bergeron-Findeisen Process

- Saturation vapor pressure for ice is lower than that for water
- Air is near saturation for water, but is supersaturated for ice
- Ice crystals/snowflakes grow by vapor deposition
- Cloud droplets may lose mass to evaporation

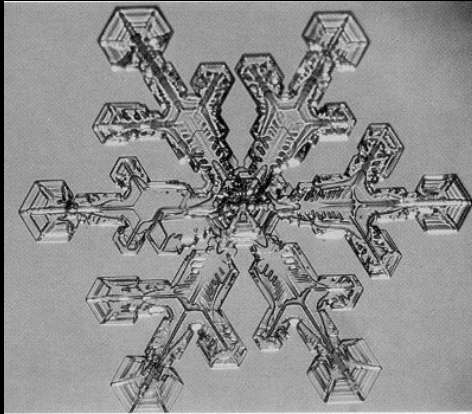
Deposition (WBF Process)



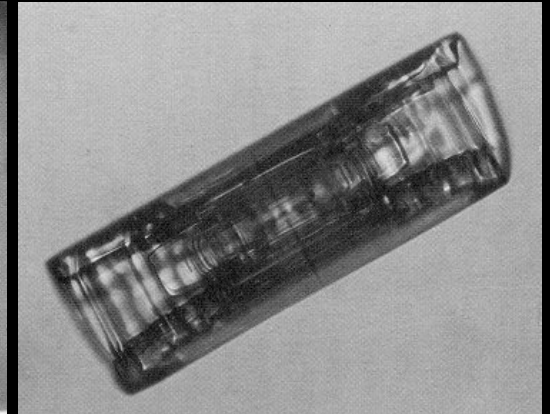
Sector Plate



Stellar Dendrite

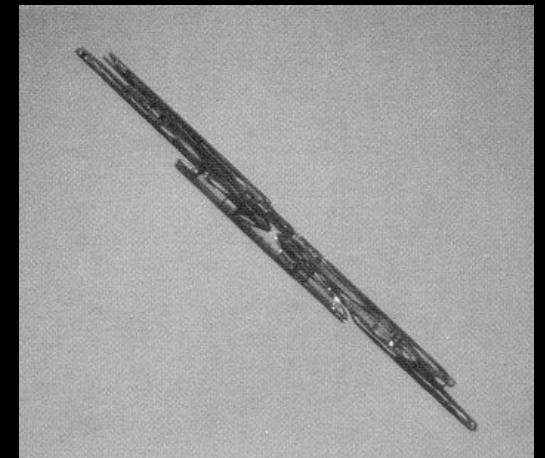


Dendritic Sector Plate



Hollow Column

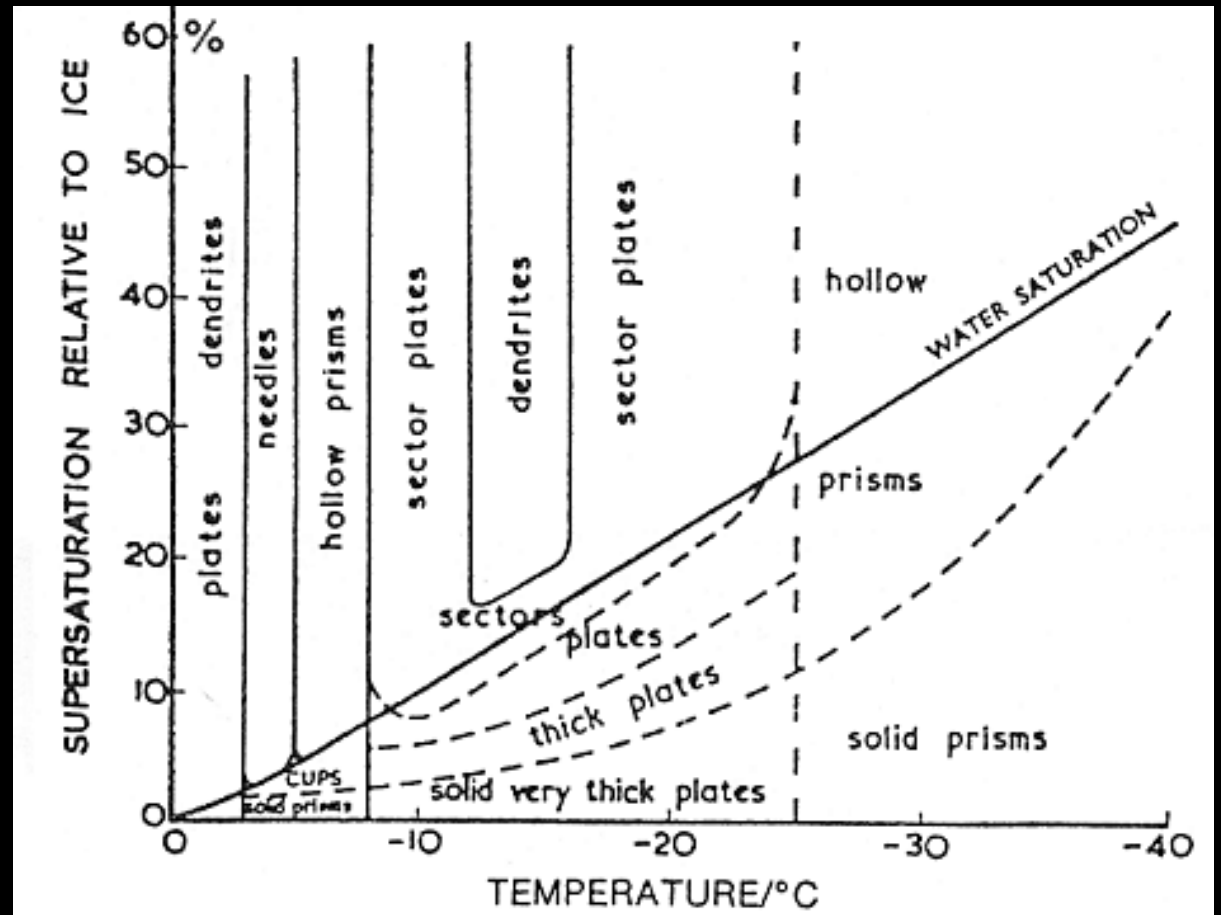
**Habits – types of ice crystal shapes
created by vapor deposition**



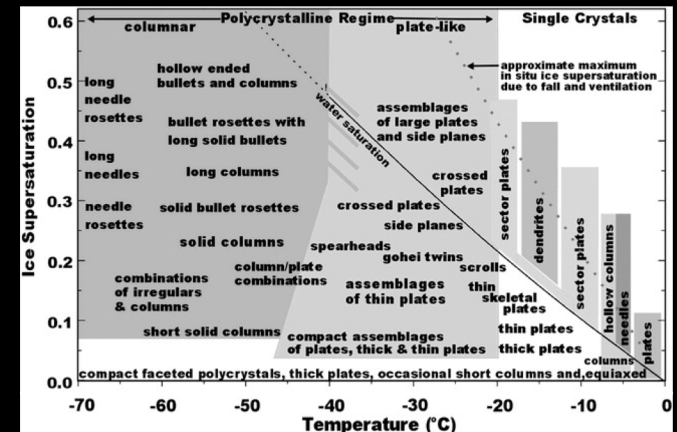
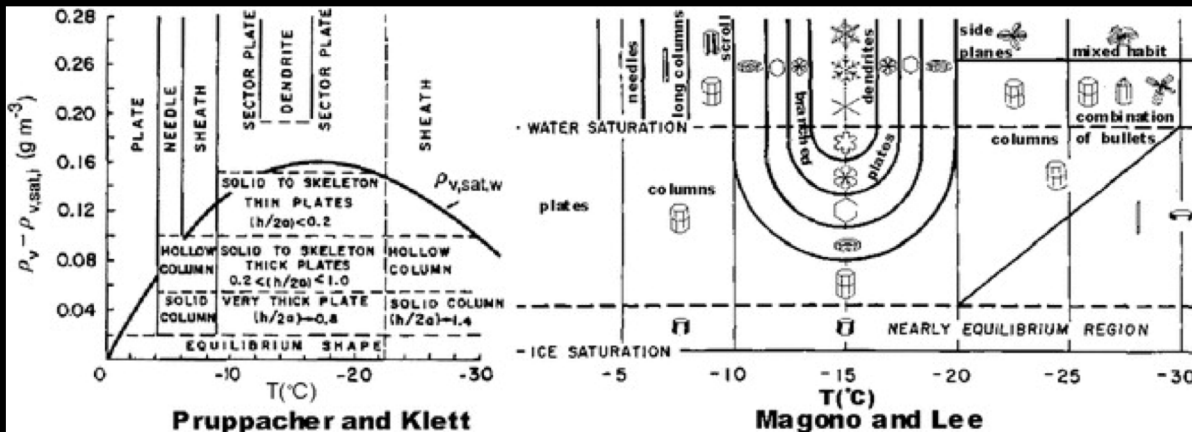
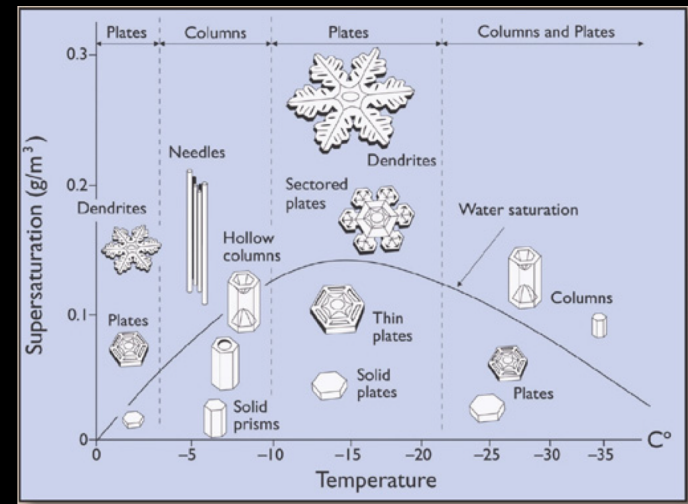
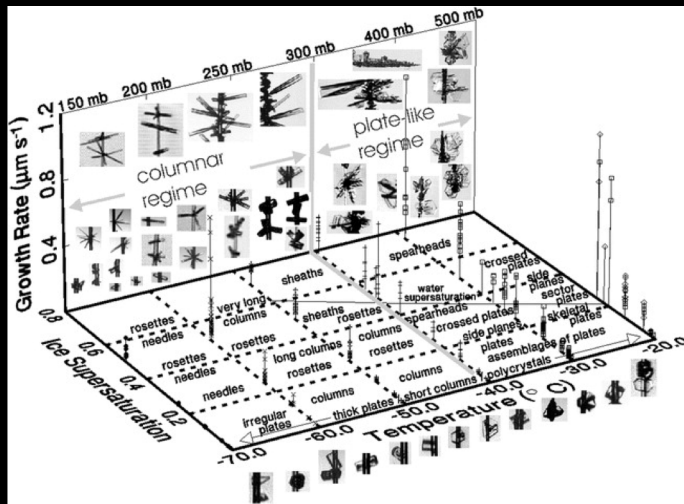
Needle

Deposition (WBF Process)

Habit is a function of temperature and supersaturation with respect to ice



More Habit Diagrams



"A review of over 70 years of ice crystal studies reveals a bewildering variety of habit diagrams"
- Bailey and Hallett (2009)

Classification Systems

Magono
And
Lee

Snow crystal classification system of Magono-Lee

N1a	Elementary needle	C1f	Hollow column	F2b	Stellar with sectorlike ends	CP3d	Plate with serrish at ends	R3c	Grasped-like with serrish extensions
N1b	Bundle of elementary needles	C1c	Solid thick plate	F2c	Dendrite with plates at ends	P7c	Stellar with spatial plates	R4a	Hexagonal grasped
N1c	Elementary sheath	C1b	Thick plate of skeletal form	P2d	Dendrite with sectorlike ends	P6d	Stellar with spatial dendrites	R4b	Lump grasped
N1d	Bundle of elementary sheaths	C1f	Scroll	F2e	Plate with simple extensions	P7a	Radiating assemblage of plates	R4c	Conifike grasped
N1e	Long solid column	C2a	Combination of bullets	P2f	Plate with sector extensions	P7b	Radiating assemblage of dendrites	I1	Ice particle
N2a	Combination of needles	C2b	Combination of columns	P2g	Plate with dendrite extensions	CF1a	Column with plates	I2	Rimed particle
N2b	Combination of sheaths	P1a	Hexagonal plate	F3a	Two branches	CF1b	Column with dendrites	I3a	Broken branch
N2c	Combination of long solid columns	P1b	Sector plate	F3b	Three branches	CF1c	Multiple capped column	I3b	Rimed broken branch
C1a	Pyramid	P1c	Broad branch	F3c	Four branches	CP2a	Bullet with plates	I4	Miscellaneous
C1b	Cup	P1d	Stellar	F4a	Broad branch with 12 branches	CP2b	Bullet with dendrites	G1	Minute column
C1c	Solid bullet	P1e	Ordinary dendrite	P4b	Dendrite with 12 branches	CP3a	Stellar with needles	G2	Germ of skeletal form
C1d	Hollow bullet	P1f	Ferulike dendrite	P5	Malformed crystal	CP3b	Stellar with columns	G3	Minute hexagonal plate
C1e	Solid column	P2a	Stellar with plates at ends	P5a	Plate with spatial branches	CP3c	Stellar with serrish at ends	G4	Minute stellar
								G5	Minute assemblage of plates
								G6	Irregular germ

International

CODE	GRAPHIC SYMBOL	TYPICAL FORMS
1		
2		
3		
4		
5		
6		
7		
8		
9		
0		

Nakaya

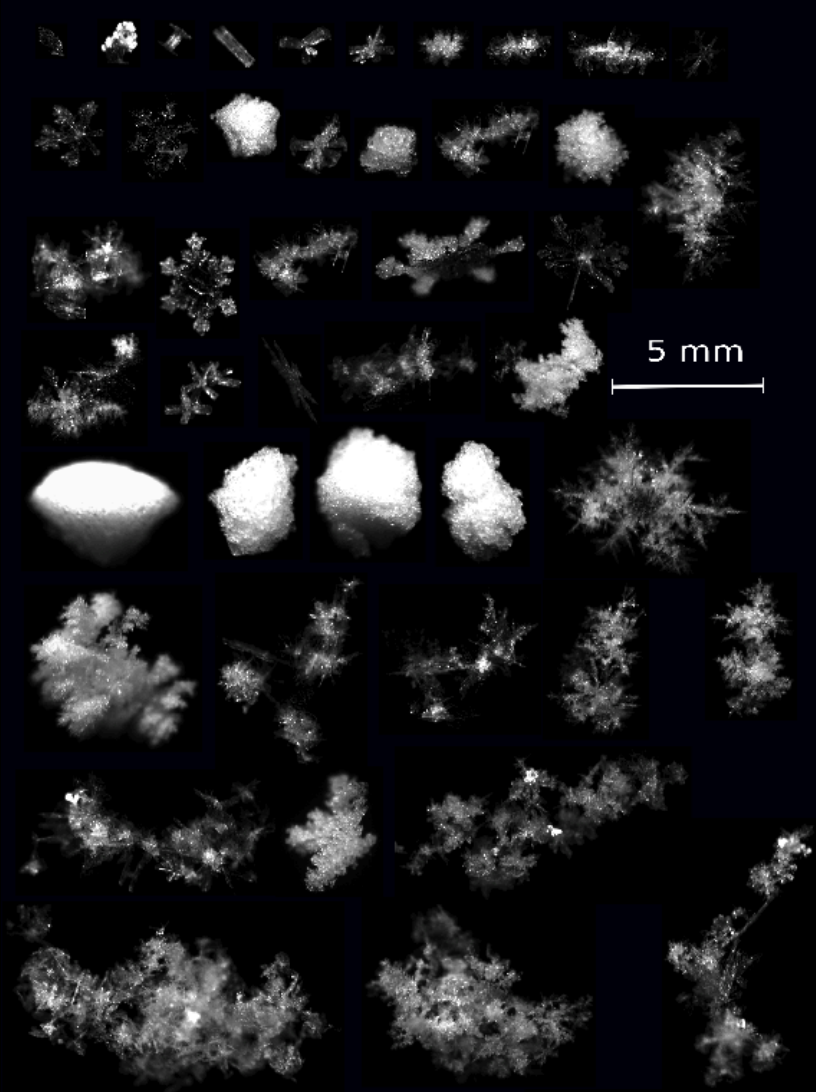
N1a	N1b	N2	C1a	C1b	C1c	C2a	C2b	
P1a	P1b	P1c	P1d	P1e	P1f	P1g	P1h	P1i
P2a	P2b	P2c	P3a	P3b	P4	P5a	P5b	
CP1a	CP1b	CP1c	CP2a	CP2b	CP3	S	I1	I2
R1	R2	R3a	R3b	R4a	R4b	R4c		

FIG. 197. General classification of snow crystals, sketches.

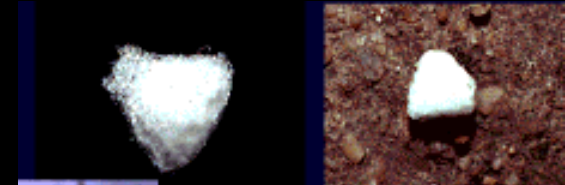
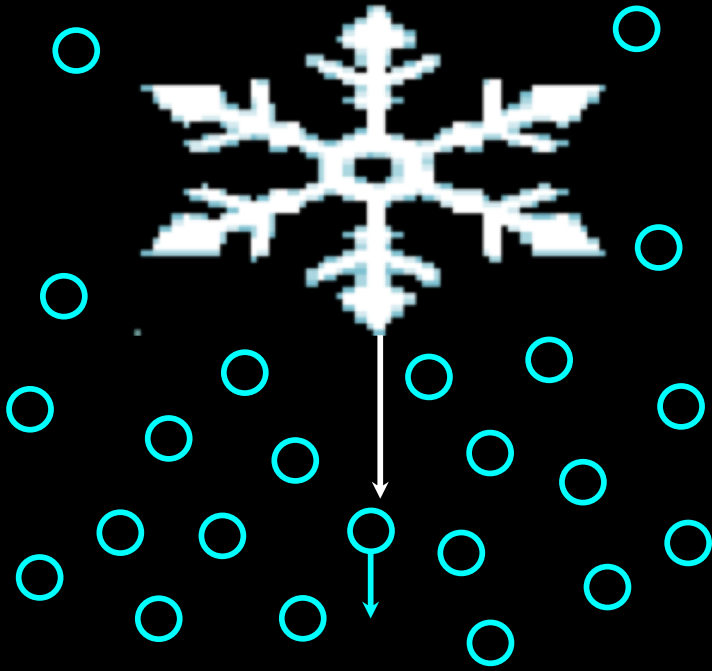
Reality

“While aesthetically appealing and offering a striking subject for photography, the fact is that most ice crystals are defective and irregular in shape”

- Bailey and Hallett (2009)



Accretion



Graupel (UCLA)



Hexagonal



Lump

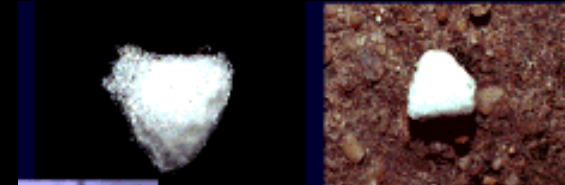
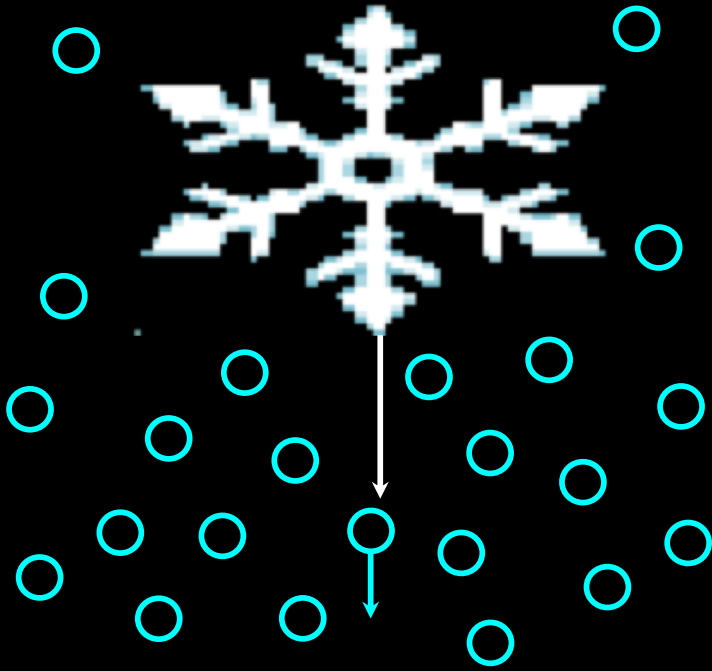


Cone

Magono and Lee (1966)

- Growth of a hydrometeor by collision with supercooled cloud drops that freeze on contact
- Graupel – Heavily rimed snow particles
 - 3 types: cone, hexagonal, lump

Accretion



Graupel (UCLA)



Hexagonal



Lump

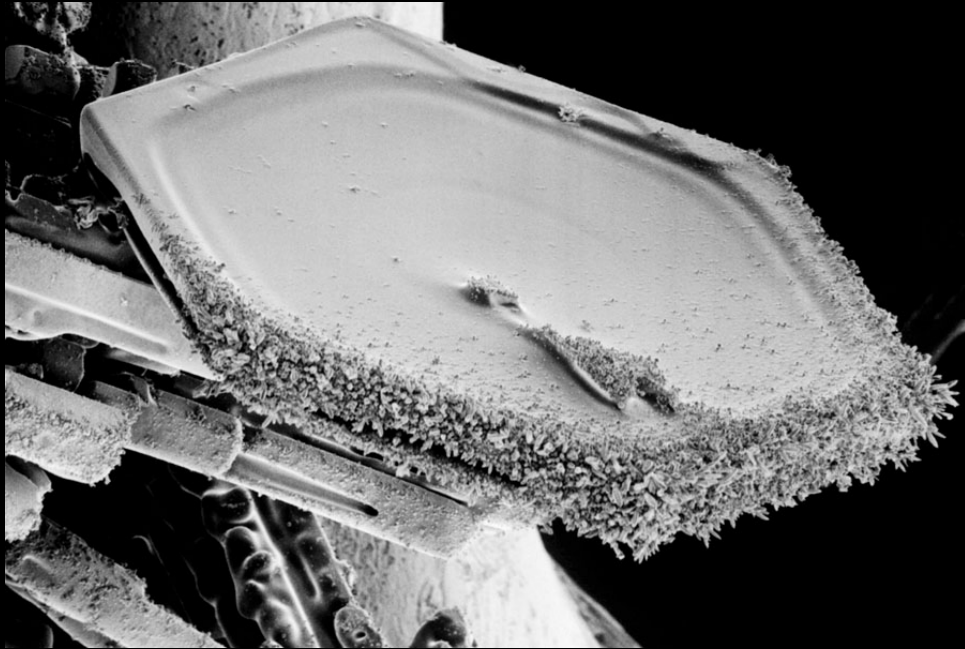


Cone

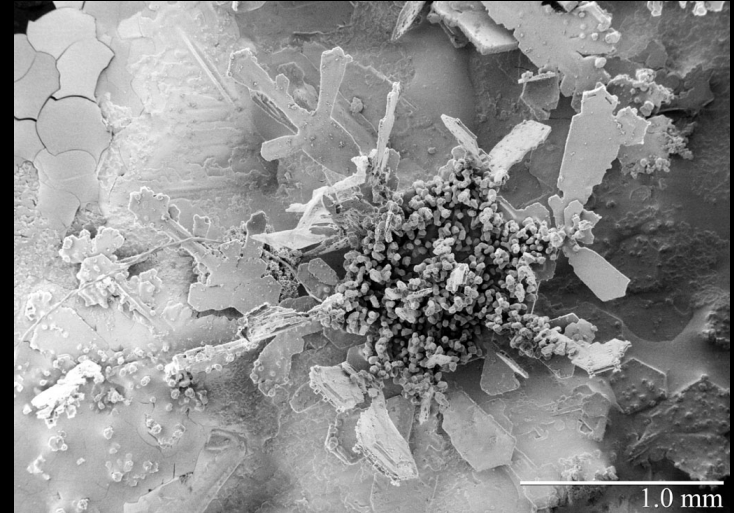
Magono and Lee (1966)

- Favored by
 - Warmer temperatures (more cloud liquid water, less ice)
 - Maritime clouds (fewer, but bigger, cloud droplets)
 - Strong vertical motion (larger cloud droplets lofted, less time for droplet cooling and ice nuclei activation)

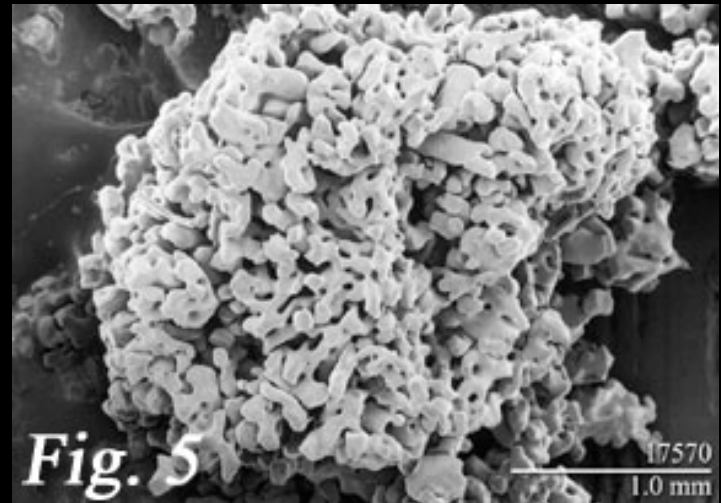
Accretion



Rimed Plate



Rimed Dendrite



Graupel

Aggregation



- Ice particles colliding and adhering with each other
- Can occur if their fall speeds are different
- Adhering is a function of crystal type and temperature
 - Dendrites tend to adhere because they become entwined
 - Plates and columns tend to rebound
 - Crystal surfaces become stickier above -5°C

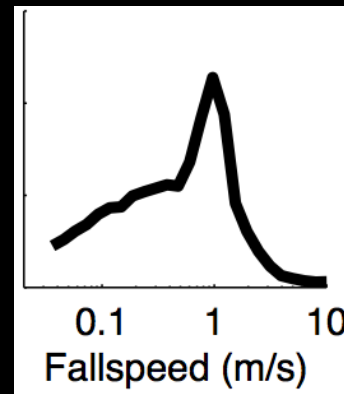
Aggregation



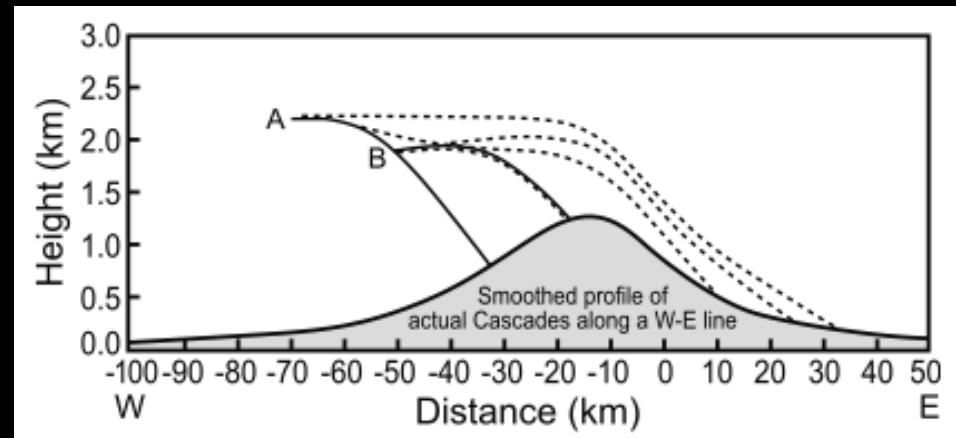
- Bigger particles
- Impact on precipitation rate is probably small
 - May impact crystal transport and fallout across mountain barriers
 - May affect mass loss from sublimation/evaporation below cloud base

Growth, Transport, & Fallout

- Growth, fallspeed, transport, and terrain scale affect precip rate and distribution
- Typical fall speeds
 - Small ice particles: $\ll 1$ m/s
 - Snow: ~ 1 m/s
 - Graupel: ~ 3 m/s
 - Rain ~ 7 m/s



Fallspeed of particles @ Alta



Solid Lines = Heavily Rimed Particles
Dashed Lines = Lightly Rimed Particles

Discussion

What evidence is there that these microphysical processes operate in Utah?

Do you have a “microphysical experience” you could share with the group?

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